Precise Carbon-13 N.M.R. Multiplicity Determination

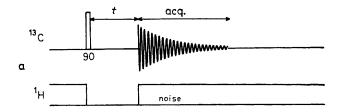
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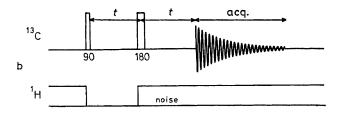
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Summary A simple excitation sequence allows fast recognition of carbon multiplicity in ¹³C n.m.r. spectra of large molecules, with high precision and sensitivity.

ROUTINE ¹³C n.m.r. analyses, namely chemical shift and carbon multiplicity determinations, are usually performed on proton noise-decoupled, and proton off-resonance decoupled spectra. However, the last technique is often

difficult to use for complex molecules. Some new methods allow better analysis of $^{13}\mathrm{C}$ n.m.r. spectra, 1,2 but are not normally usable with routine spectrometers, as they use software or hardware alterations (e.g. decoupler phase-shift). We present in this communication a method allowing easy identification of quaternary, CH, CH₂, and CH₃ carbons in $^{13}\mathrm{C}$ n.m.r. spectra, with a sensitivity similar to that of proton noise-decoupled techniques. The basic features of this multiplicity separation method have already been used by Campbell et al.³ for $^{1}\mathrm{H}$ n.m.r. studies on proteins.





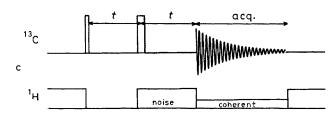
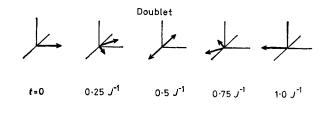


FIGURE 1. Pulse sequences. (a) Simple sequence of multiplicity separation; (b) modified sequence to avoid phase distortion; (c) sequence for observation of off-resonance multiplets.

The fundamental experiment is depicted in Figure 1a. The sequence is a standard proton-decoupled carbon sequence except that during a short period t before the acquisition the proton decoupler is turned off. Analysis of carbon magnetization in the rotating frame (Figure 2) clearly shows that a variety of information can be gained depending upon the choice of the t value. If t takes the value $1/J(^{1}H^{-13}C)$ the CH and CH₃ signals are inverted, whilst the CH₂ and quaternary carbon signals are positive.

In general, $\rm sp^3$ carbons are coupled to protons with a coupling constant J of about 125 Hz, and selecting a value of 8 ms for t allows complete recognition of the parity of all carbon peaks. However, the delay introduced before acquisition, leading to a truncated free-induction decay



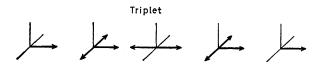


FIGURE 2. Behaviour of doublet and triplet magnetization with the same coupling constant J, after a 90° excitation pulse, in the rotating frame.

(F.I.D.), produces a severe frequency-dependent phase error and line-shape distortion. Sequence 1b (Figure 1) allows elimination of this by utilisation of a spin echo.



FIGURE 3. Upfield part of the ¹³C (62.86 MHz) n.m.r. spectrum of emericide (1). Phase corrections have been made so that CH and CH₃ resonances appear as positive, and quaternary and CH₂ carbons as negative signals. *CDCl₃ resonance.

A typical example is given in Figure 3 on emericide (1), an ionophorous antibiotic. This example shows that the CH and CH₂ resonances, separated by only 10 Hz, are well resolved. It gives a better selectivity and sensitivity than classical off-resonance experiments and has a broader field of application than selective T_1 experiments. The

sequence 1b produces positive and negative peaks depending upon parity of the number of protons, while the simple spin echo with continuous decoupling produces only positive peaks. It is therefore evident that one sequence 1b followed by one fully decoupled spin echo gives an F.I.D. where only signals due to quaternary and CH₂ carbons are observed. In the same way, one sequence 1b followed by one inverted decoupled spin echo produces a spectrum of signals of even parity only. In practice, separating even and odd parity peaks is especially useful when using the

modified sequence 1c (Figure 1) where the acquisition occurs under proton low-power single-frequency decoupling.

In general, the efficiency of low-power heteronuclear decoupling applied to the identification of CH2 carbons (1H-13C chemical shift correlation) is very low if CH2 resonances are coincident with those of CH and CH3 carbons, and if both protons have different chemical shifts; the carbon signal often appears as either a broad quadruplet. if the low power decoupling is off-resonance, or a doublet if one of the two protons is fully decoupled. Both patterns have insufficient amplitude to be clearly visible in the middle of the CH or CH3 multiplets. However, use of sequence 1c applied in alternation with decoupled spinecho leads to the observation of CH2 multiplets only, and allows proton-carbon chemical shift correlations to be made in some difficult cases.

During the preparation of this manuscript a different method for separating carbon peaks according to their multiplicity was described.7 It is based on a different extension of the spin-echo sequence; and seems to be more difficult to use on routine spectrometers.

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